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## Key indicators

Single-crystal X-ray study  
 $T = 273$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å  
Disorder in solvent or counterion  
 $R$  factor = 0.049  
 $wR$  factor = 0.140  
Data-to-parameter ratio = 12.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.Poly[[bis[ $\mu$ -1-ethyl-6-fluoro-4-oxo-7-(1-piperazinyl)-1,4-dihydro-1,8-naphthyridine-3-carboxylato]-cobalt(II)] dihydrate]

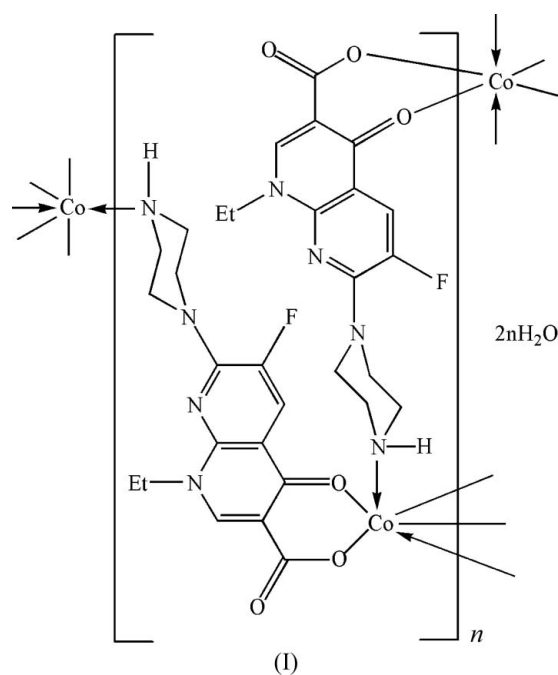
In the title compound,  $\{[\text{Co}(\text{C}_{15}\text{H}_{16}\text{FN}_4\text{O}_3)_2] \cdot 2\text{H}_2\text{O}\}_n$ , the  $\text{Co}^{\text{II}}$  atom (site symmetry  $\bar{1}$ ) exists in a distorted *trans*- $\text{CoN}_2\text{O}_4$  octahedral geometry that is defined by two monodentate *N*-bonded and two bidentate *O,O*-bonded 1-ethyl-6-fluoro-4-oxo-7-(1-piperazine)-1,4-dihydro-1,8-naphthyridine-3-carboxylate (enox) monoanions. The extended two-dimensional structure is a square grid.

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## Comment

Enoxacin (H-Enox, 1-ethyl-6-fluoro-4-oxo-7-piperazine-1,4-dihydro-1,8-naphthyridine-3-carboxylic acid) is a member of the class of quinolones that is used to treat infections (Mizuki *et al.*, 1996). Manganese(II) and cadmium(II) derivatives of enox have been reported (Yu *et al.*, 2005; Zhang *et al.*, 2006). The title cobalt(II) derivative, (I), a two-dimensional coordination polymer in which the anion acts in a bridging mode, is reported here (Fig. 1).



The  $\text{Co}^{\text{II}}$  atom (site symmetry  $\bar{1}$ ) in (I) is coordinated (Table 1) by four O atoms and two N atoms from four enoxacin ligands (two *N*-monodentate and two *O,O*-bidentate), forming an approximate square grid (Fig. 2) propagating in  $(10\bar{2})$ . The disordered uncoordinated water molecules occupy cavities within the grid. An  $\text{N}-\text{H} \cdots \text{O}$  hydrogen bond (Table 2) to carboxylate atom O2 (not bonded to Co) completes the structure.

## Experimental

A mixture of  $\text{Co}(\text{CH}_3\text{COO})_2 \cdot 4\text{H}_2\text{O}$  (0.062 g, 0.25 mmol), Henox (0.13 g, 0.5 mmol) and water (12 ml) was stirred for 30 min in air. The mixture was then transferred to a 23 ml Teflon-lined hydrothermal bomb. The bomb was kept at 423 K for 72 h under autogenous pressure. Pink single crystals of (I) suitable for X-ray analysis were obtained from the reaction mixture after cooling.

### Crystal data

$[\text{Co}(\text{C}_{15}\text{H}_{16}\text{FN}_4\text{O}_3)_2] \cdot 2\text{H}_2\text{O}$	$Z = 2$
$M_r = 729.57$	$D_x = 1.470 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 5.9422(4) \text{ \AA}$	$\mu = 0.59 \text{ mm}^{-1}$
$b = 21.4433(16) \text{ \AA}$	$T = 273(2) \text{ K}$
$c = 13.147(1) \text{ \AA}$	Prism, pink
$\beta = 100.290(2)^\circ$	$0.32 \times 0.26 \times 0.18 \text{ mm}$
$V = 1648.3(2) \text{ \AA}^3$	

### Data collection

Bruker SMART CCD diffractometer	8212 measured reflections
$\omega$ scans	2921 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	2204 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.833$ , $T_{\max} = 0.901$	$R_{\text{int}} = 0.049$
	$\theta_{\text{max}} = 25.1^\circ$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0722P)^2 + 0.8753P]$
$R[F^2 > 2\sigma(F^2)] = 0.049$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.140$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.58 \text{ e \AA}^{-3}$
2921 reflections	$\Delta\rho_{\text{min}} = -0.28 \text{ e \AA}^{-3}$
231 parameters	
H atoms treated by a mixture of independent and constrained refinement	

**Table 1**

Selected bond lengths ( $\text{\AA}$ ).

Co1—O1	2.047 (2)	Co1—N4 <sup>i</sup>	2.254 (2)
Co1—O3	2.0718 (19)		

Symmetry code: (i)  $-x, y + \frac{1}{2}, -z + \frac{1}{2}$ .

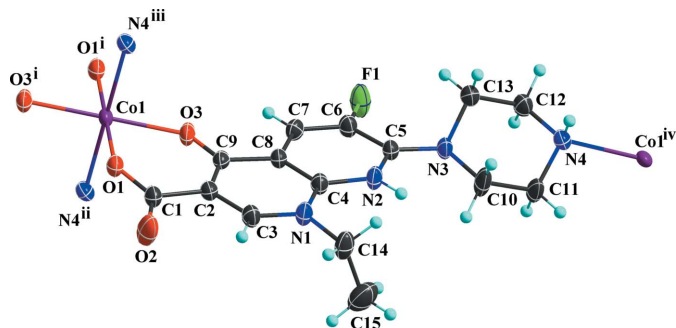
**Table 2**

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D\text{—H}\cdots A$	$D\text{—H}$	$\text{H}\cdots A$	$D\cdots A$	$D\text{—H}\cdots A$
$\text{N4—H4}\cdots\text{O2}^{\text{ii}}$	0.86 (3)	2.29 (3)	3.124 (4)	163 (2)

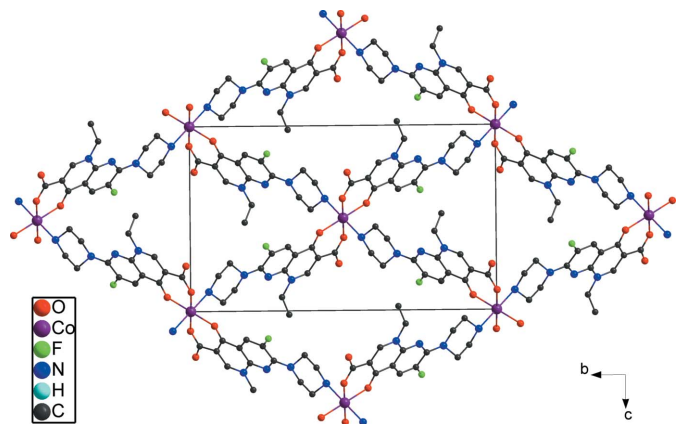
Symmetry code: (ii)  $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$ .

The carbon-bound H atoms were positioned geometrically ( $C\text{—H} = 0.93\text{--}0.97 \text{ \AA}$ ) and refined as riding, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . The N-bound H atom was located in a difference map and refined with a distance restraint of  $0.86(1) \text{ \AA}$  and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$ . The uncoordinated water molecule is disordered over two adjacent  $[\text{O}\cdots\text{O} = 2.138(6) \text{ \AA}]$  positions; the fractional site occupancies refined to  $0.557(5):0.443(5)$  (sum constrained to unity). The water H atoms could not be placed due to this disorder.



**Figure 1**

The asymmetric unit of (I), extended to show the Co coordination, showing 50% displacement ellipsoids (H atoms and water molecule O atoms have been omitted for clarity). [Symmetry codes: (i)  $1 - x, 1 - y, 1 - z$ ; (ii)  $-x, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (iii)  $x + 1, -y + \frac{1}{2}, z + \frac{1}{2}$ .]



**Figure 2**

A view of part of a two-dimensional polymeric sheet in (I) showing the square-grid connectivity (H atoms and water molecule O atoms have been omitted for clarity).

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINTE* (Bruker, 1998); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1998); software used to prepare material for publication: *SHELXTL*.

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